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Acta Cryst. (1998). C54, 1143-1144

$(1\alpha,3\alpha,5\alpha)$ -1,3,5-Trimethyl-1,3,5-cyclohexanetricarboxylic Acid Acetonitrile Solvate

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(Received 30 September 1997; accepted 15 January 1998)

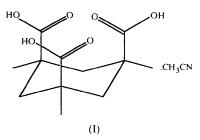
Abstract

In the title compound, $C_{12}H_{18}O_6.CH_3CN$, both components have crystallographically imposed C_3 symmetry. The hydrogen bonds between the three carboxylic acid groups of $(1\alpha,3\alpha,5\alpha)$ -1,3,5-trimethyl-1,3,5-cyclohexanetricarboxylic acid (Kemp's triacid) and three other triacid moieties make a three-dimensional hydrogen-bonding network, producing large intermolecular cavities.

Comment

During a chemical study on the interactions between metal ions and several kinds of $(1\alpha, 3\alpha, 5\alpha)$ -1,3,5-trimethyl-1,3,5-cyclohexanetricarboxylic acid (Kemp's

triacid) (Hirose *et al.*, 1995; Baldwin *et al.*, 1996), the crystal structures of the acid and an adduct were determined. The structure determined for the acid was disordered [trigonal, a = 25.152 (2), c = 12.592 (2) Å, Z = 18, $R\bar{3}$] and is yet to be published. This paper reports the acetonitrile adduct, (I), with the cyclohexane ring having a chair conformation. There is a hydrogen-bonding system between O11 and a neighbouring O10 atom at (1 - x, 1 - y, -z) [O···O = 2.651 (3) Å and O11—H11···O10 = 175.4 (3)°]. There are no links between the acetonitrile and the triacid.



For the compound under investigation, each carboxylic acid group of each Kemp's triacid formed intermolecular hydrogen bonds with a centrosymmetrically related neighbour. A three-dimensional interconnecting network (Fig. 2), formed by hydrogen bonds, links neighbouring acid groups to give six-membered sets of Kemp's acids, in alternating hydrogen-bond links above and below the ab plane around the lattice points (0,0,0), $(\frac{2}{3}, \frac{1}{3}, \frac{1}{3})$ and $(\frac{1}{3}, \frac{2}{3}, \frac{2}{3})$. The acetonitrile lies outside this framework, in a void of 172 Å³. Corey-Pauling-Kendrew (CPK) space-filling model studies showed that large intermolecular cavities would be formed within the crystal structure if this compound kept its C_3 symmetry. The introduction of acetonitrile is a stabilizing influence, as it would fill this void. The structure of the disodium tetrahydrate has also been determined (Bencini et al., 1994).

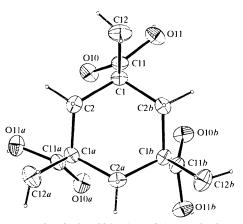


Fig. 1. *PLATON*96 (Spek, 1996) view of the $(1\alpha,3\alpha,5\alpha)$ -1,3,5-trimethyl-1,3,5-cyclohexanetricarboxylic acid of the title compound, with displacement ellipsoids at the 20% probability level. [Symmetry codes: (a) 1 - y, x - y, z; (b) 1 - x + y, 1 - x, z.]

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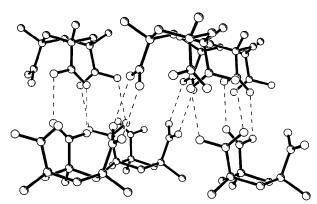


Fig. 2. Part of the three-dimensional interconnecting hydrogen-bonded network of Kemp's acid molecules. The acetonitrile molecules, which lie outside this framework, have been omitted.

Experimental

Crystals of (I) were grown by recrystallization of $(1\alpha, 3\alpha, 5\alpha)$ -1,3,5-trimethyl-1,3,5-cyclohexanetricarboxylic acid from acetonitrile/water solution (1:1 v/v). The unadducted compound was first prepared by Kemp & Petrakis (1981). Its structure was confirmed in a space group without threefold crystallographic symmetry by Rebek *et al.* (1985) [a = 8.471 (2), b =12.138 (3), c = 12.902 (3) Å, $\beta = 101.8$ (2)°, $P2_1/n$] and by Chan *et al.* (1991) [a = 8.465 (2), b = 12.136 (2), c =12.892 (2) Å, $\beta = 101.78$ (1)°, $P2_1/n$].

Crystal data

$C_{12}H_{18}O_6.C_2H_3N$	Mo $K\alpha$ radiation
$M_r = 299.32$	$\lambda = 0.71073 \text{ Å}$
Trigonal	Cell parameters from 25
<i>R</i> 3	reflections
a = 8.972(1) Å	$\theta = 10 - 14^{\circ}$
c = 34.15(2) Å	$\mu = 0.098 \text{ mm}^{-1}$
$V = 2380.7 (14) \text{ Å}^3$	T = 293 (2) K
Z = 6	Large block
$D_x = 1.253 \text{ Mg m}^{-3}$	$0.47 \times 0.47 \times 0.40$ mm
D_m not measured	Colourless

Data collection

MicroVAX-controlled
CAD-4 diffractometer
$2\theta/\omega$ scans
Absorption correction: none
3043 measured reflections
933 independent reflections
641 reflections with
$I > 2\sigma(I)$

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.050$ $wR(F^2) = 0.152$ S = 1.096933 reflections 92 parameters

$$R_{int} = 0.045$$

$$\theta_{max} = 25^{\circ}$$

$$h = -10 \rightarrow 10$$

$$k = -10 \rightarrow 9$$

$$l = -40 \rightarrow 40$$

3 standard reflections
frequency: 120 min
intensity decay: none

 $w = \frac{1}{[\sigma^2(F_o^2) + (0.047P)^2 + 6.17P]}$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} = 0.001$ $\Delta\rho_{max} = 0.15$ e Å⁻³

 $\Delta \rho_{\rm min} = -0.13 \ {\rm e} \ {\rm \AA}^{-3}$

H atoms refined by a mixture of independent and constrained refinement Extinction correction: none Scattering factors from International Tables for Crystallography (Vol. C)

The title structure was solved using direct methods.

Data collection: *SDP* (Frenz, 1985). Cell refinemert: *SDP*. Data reduction: *Xtal3.0* (Hall & Stewart, 1990). Program(s) used to solve structure: *SHELXS86* (Sheldrick, 1990). Program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997). Molecular graphics: *PLATON96* (Spek, 1996) and *SHELXTL/PC* (Sheldrick, 1994). Software used to prepare material for publication: *SHELXL97*.

The authors wish to thank the Australian Research Council for the purchase of the CAD-4 diffractometer, and the University of Queensland and National Institute of Materials and Chemical Research for financial support.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: BM1202). Services for accessing these data are described at the back of the journal.

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